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Camilla Rothe

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only downside to using skeleton equations as opposed to balanced equations is that skeleton equations don't tell you the quantity of stuff that reacts

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web there's 2 on the left and 1 on the right so we need to change the coefficient of NH_3 to 2 now we have $\text{N}_2 + \text{H}_2 \rightarrow 2 \text{ NH}_3$ total the atoms up again on the left there is 2 N and 2 H still

on the right there is 2 N and 6 H now so now all we need to do is make the left side have 6 H in total

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the carbon atoms first 5 use a coefficient to balance the
single carbon atom add a coefficient to the single carbon

atom on the right of the equation to balance it with the 3
carbon atoms on the left of the equation c 3 h 8 o 2 h 2 o 3 co
2