

Inspiration Is For Amateurs

K Morrison

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energy to support your answer answer
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energy than the reactants b on the
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if a catalyst is added 46 given the
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web when the reactants have more

potential energy than the products have
and the extra energy is released to the
surroundings it is called activation
energy the height of the hill on a
potential energy diagram is a measure
of the endothermic the reaction above
can be categorized as

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web this is like a one dimensional
system whose mechanical energy e is a
constant and whose potential energy
with respect to zero energy at zero
displacement from the spring s
unstretched length x_0 is $u = \frac{1}{2} kx^2$
2 figure 8 5 2 8 5 2 a a glider between
springs on an air track is an example of
a horizontal mass spring system

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web 1 if the pe of the reactants is 430kj
and the pe of the products is 190kj
what is the Δh 2 if the pe of the
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will be needed to be added for the
reaction to proceed transcribed image
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diagram shows the total potential
energy of a reacting system as the
reaction proceeds a in an endothermic
reaction the energy of the products is
greater than the energy of the
reactants and Δh is positive b in an
exothermic reaction the energy of the
products is lower than the energy of the
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rate of a chemical reaction $a b c d$ is
given in first approximation by $v = k_v a^b$
where $k_v = k e^{-E_a/RT}$ where k depends by
many factors as solvent geometry type
of reaction E_a is the activation energy R
is the universal constant of gases T is the
absolute temperature

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web aug 8 2022 the energy changes that occur during a chemical reaction can be shown in a diagram called a potential energy diagram or sometimes called a reaction progress curve a potential energy diagram shows the change in potential energy of a system as reactants are converted into products the figure below shows basic potential energy

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web key points potential energy is energy that has the potential to become another form of energy an object s potential energy depends on its physical properties and position in a system potential energy comes in many

forms such as gravitational potential energy due to an object s mass and position in a gravitational field answered use the potential energy diagram shown bartleby - Jul 21 2022 web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate Δh for the reaction and label it with the correct value see attached image **why is activation energy drawn in a potential energy diagram in reactions** - Dec 26 2022

web aug 16 2019 1 answer sorted by 4 the activation energy is not the average kinetic energy that the colliding particles need in order to successfully react instead it is the energy barrier between reactants and products often the energy required to cross the barrier comes from the kinetic energy of collisions

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web all questions are answered herein chemistry tutorial 9 02b potential energy diagrams how to draw and label

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 your drawing with the following letters
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 b pe of reactants e heat of reaction c pe
 of the activated complex potential
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 is the
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