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Patrick Vollmar

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Quantum Electrodynamics of Photosynthesis Artur Braun, 2020-10-12 This book uses an array of different approaches to describe photosynthesis, ranging from the subjectivity of human perception to the mathematical rigour of quantum electrodynamics. This interdisciplinary work draws from fields as diverse as astronomy, agriculture, classical and quantum optics, and biology in order to explain the working principles of photosynthesis in plants and cyanobacteria.

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we doomed to repeat history, or can we adapt to safeguard our future?

Innovations in Chemical Biology Bilge Sener, 2008-11-23 This book includes 49 chapters presented as plenary, invited lectures and posters at the conference. Six plenary lectures have been published in an issue of Pure and Applied Chemistry, Vol. 79, No. 12, 2007; the titles of these presentations are given as an Annex at the end of the book. I thank all contributors for the preparation of their presentations. It is sad to report that Professor Hitoshi Ohtaki, one of the founders of the Eurasia conferences and contributors passed away on November 5, 2006. Professor Ohtaki enthusiastically promoted international cooperation and took it upon himself to publicize Japanese science to the wider world. His contribution in this book will serve as a memorable contribution to that goal. He will be missed by all of us. This book is dedicated to his memory. Professor Dr. Bilge Sener Editor Memorial Tribute to Professor Dr. Hitoshi Ohtaki Curriculum Vitae of Hitoshi Ohtaki Date of Birth September 16, 1932 Place of Birth Tokyo, Japan Date of Decease November 5, 2006 (at the age of 74) Address 3-9-406 Namiki-2-chome, Kanazawa-ku, Yokohama, Japan Institution Chair Professor of The Research Organization of Science and Engineering, Ritsumeikan University Guest Professor of Yokohama City University Education Bachelor of Science, Nagoya University, 1955 Master of Science, Nagoya University, 1957 Doctor of Science, Nagoya University, 1961 ix x Memorial Tribute to Professor Dr.

Chemical Elements,

The Nature of the Chemical Bond Linus Pauling, 1960 Resonance and the chemical bond. The partial ionic character of covalent bonds and the relative electronegativity of atoms. The directed bond. Bond strengths and bond angles. The magnetic criterion for bond type. The resonance of molecules among several valence-bond structures. Interatomic distances and their relation to the structure of molecules and crystals. Types of resonance in molecules. The structure of molecules and complex ions involving bonds with partial double-bond character. The one-electron bond and the three-electron bond. The hydrogen bond. The sizes of ions and the structure of ionic crystals. The metallic bond. A summarizing discussion of resonance and its significance for chemistry.

Journal and Proceedings of the Royal Society of New South Wales Royal Society of New South Wales, 1876 Includes list of members.

The Hutchinson Encyclopedia of Science, 1998 Whether it's the planet, the petri dish or the platform, this reference attempts to shed some light on the subject. It includes over 10,000 entries and many articles describing discoveries and ground-breaking experiments and is fully cross-referenced.

Chemical Bonds Phillip Manning, 2009 Modern life is made up of a mind-boggling array of materials. A simple drinking cup, for example, might be made of Styrofoam, paper, or glass, depending on the drinker's needs at the moment. Home storage cabinets can be made of metal, wood, or plastic. Space shuttles are assembled from silicon, steel, and hundreds of

other materials. All of these items owe their properties to the chemical bonds between the atoms that make up the substance. Chemical Bonds examines the nature of the chemical bonds, answering fundamental questions about how they form, how they are broken, and how they help define life as we know it.

Quarterly Journal of the Chemical Society of London Chemical Society (Great Britain),1965

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