Magnesium And Hydrogen Chloride

K Morrison

A Comprehensive Treatise on Inorganic and Theoretical Chemistry Joseph William Mellor, 1922 The Manufacture of Hydrochloric Acid and Saltcake Alex Charles Cumming, 1923

The Recovery of Magnesium Oxide and Hydrogen Chloride from Magnesium Chloride Brines and Molten Salt Hydrates Jan Simon Christiaan De Bakker, 2011 Hydrochloric acid leaching of saprolite nickel ores has been proposed as an effective means of recovering nickel and cobalt. However, the leach produces a concentrated brine of magnesium chloride which must be hydrolyzed to recover the HCl lixiviant. The processing of carnallite similarly produces a concentrated MgCl2 brine; converting this brine into HCl and MgO provides an attractive way of adding value while effectively disposing of this waste product. Direct pyrohydrolysis of magnesium chloride brines by the reaction, MgCl2,a + H2Oa [produces]MgOs + 2HClg is energy-intensive as large volumes of water must be evaporated. The energy cost is high, and the HCl stream produced is limited to approximately 20 wt% HCl. This thesis explores alternative methods of obtaining HCl from aqueous magnesium chloride solutions. Two methods are considered: the hydrolysis, under autogenous pressure, of concentrated MqCl2 molten salt hydrates; and the precipitation of magnesium hydroxychloride compounds such as 2MqO·MqCl2·6H2O and 3MgO·MgCl2·11H2O, which are subsequently decomposed at high temperature. Considerable experimental difficulties were encountered in studying pressure hydrolysis of molten salt hydrates, despite extensive equipment modifications. Ultimately, the work moved on to precipitation and decomposition of hydroxychlorides. This was found to bear promise, and conceptual flowsheets based on these reactions are presented. A phase stability diagram giving the areas of predominance of the different hydroxychloride phases is presented, and fundamental thermochemical data are derived. The results of a kinetic study on magnesium hydroxychloride thermal decomposition are also presented.

<u>Supplement to Mellor's Comprehensive Treatise on Inorganic and Theoretical Chemistry: suppl. 3. K, Rb, Cs, Fr</u> Joseph William Mellor,1922

Mechanism for the Reaction Magnesium-hydrogen Chloride-ammonia for the Preparation of Large Quantities of Pure, Anhydrous Magnesium Chloride James Edward Hurse, 1961

<u>Chemistry For Middle Class-8 (Book-III)</u> R.P. Rana, The multi-coloured edition of the books has been revised and updated in a lucid language. Project work at the end of each chapter lays emphasis on the learning by doing methodology. Unit Test

Paper and Model Test Paper have been included.

Chemistry insights 'O' level Rex M. Heyworth, 2007

The Elements of Chemistry Paul Caspar Freer, 1895

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Journal of the Chemical Society ,1921

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Reviewing Magnesium And Hydrogen Chloride: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "Magnesium And Hydrogen Chloride," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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