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 would result in two moles of bismuth a total of 417 96g
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 would result in three moles of tellurium a total of 382 8g add
 the mass of bismuth and the mass of tellurium per mole to
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 through covalent bonds in salts it is held together with ionic
 bonds these are the two types of bonds out of which every
 compound is made of 2 example of compounds example of
 compounds includes water h₂o hydrogen peroxide h₂o₂ etc
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 types of chemical compounds molecular and ionic methane
 and water are composed of molecules that is they are
 molecular compounds sodium chloride on the other hand
 contains ions it is

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 ions but instead consist solely of discrete neutral molecules
 these molecular compounds covalent compounds result when
 atoms share rather than transfer gain or lose electrons
 covalent bonding is an important and extensive concept in
 chemistry and it will be treated in considerable

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 of the following compounds state whether it is ionic or
 covalent if it is ionic write the symbols for the ions involved a
 nf 3 b bao c nh 4 2 co 3 d sr h 2 po 4 2 e ibr f na 2 o answer a
 answer b answer c answer d answer e answer f problem 4 3

1 4 4 3 1 4

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iron atoms sulfur is another element it contains only sulfur
atoms and nothing else so it is pure when we mix two
different pure

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compound co h 2s sf 6 n 2o 2 strategy first you have to know
how to name a molecular compound to name a compound its
the prefix name of first element prefix and name of the
second element with the suffix ide the prefixes are as follows
mono 1 di 2 tri 3 tetra 4

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co₂ co 2 these compounds are very different from ionic
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each of the molecules a ch 4 b chcl 3 c c 12 h 10 o 6 d ch 3
ch 2 ch 2 ch 2 ch 3 answer a answer b answer c answer d
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each of the following a p 4 b h 2 o c ca no 3 2 d ch 3 co 2 h
acetic acid