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Guillermo Garzón G.

Principios de química Richard E. Dickerson, 1992-11 Esta tercera edición de Principios de Química, se ha proyectado para ser utilizada en un curso universitario de Química general, que debe servir tanto como una visión de la Química para los especialistas, como de una buena base para los estudios posteriores de las disciplinas de Química. Por consiguiente, hay varios capítulos que pretenden introducir las diferentes áreas de la Química, incluida la inorgánica, nuclear, orgánica y Bioquímica, y se intenta, a lo largo de todo el libro, colocar a la Química en su marco histórico y cultural. Al mismo tiempo, se presentan los aspectos cuantitativos de la Química de forma consecuente con su importancia, de manera que resulte fácil apoyarse en ellos en los cursos posteriores.

Principios básicos de química Harry B. Gray, Gilbert P. Jr. Haight, 2021-07-01 Este texto es una respuesta al reto que significa presentar una relación innovadora y moderna de los principios de la Química. El rápido desarrollo de los últimos cursos preuniversitarios está creando presiones sin precedentes que obligan a cambiar la enseñanza en el curso de iniciación. En este nivel se necesitan urgentemente aproximaciones al estudio de los campos más avanzados de la Química, con una presentación puesta al día. Además del texto propiamente dicho, existe la Guía para el profesor.

Estructura atómica y enlace químico Jaime Casabó Gispert, 1996 Esta obra pretende ser un libro de texto destinado a los estudiantes de primer ciclo de las Facultades de Química, Ciencias y Universidades Politécnicas, que tengan en sus planes de estudio una o varias asignaturas dedicadas al estudio del enlace químico.

Manual de Experimentos de Laboratorio Para Química i Y ii ,

Química: la Ciencia Central Theodore L. Brown, H. Eugene LeMay, Bruce Edward Bursten, Julia R. Burdge, 2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish. Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear

Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry. For anyone interested in learning more about chemistry.

Teoría y problemas de fundamentos de química general Guillermo Garzón G., 1984

Introducción a la química Leo J. Malone, 2006 Química : materia, cambios y energía - Mediciones en química - El átomo, la estructura de la materia y las reacciones nucleares - El arreglo de los electrones en el átomo - La naturaleza periódica de los elementos - La naturaleza del enlace químico - La nomenclatura de los compuestos - Relaciones cuantitativas : la mol - Reacciones químicas : ecuaciones y cantidades - El estado gaseoso de la materia - Agua : los estados líquidos y sólidos - Soluciones acuosas - Ácidos, bases y sales - Reacciones de oxidación-reducción - Cinética y equilibrio químico - Introducción a la química orgánica.

Ibérica , 1919

Boletín Centro de Documentación Científica y Técnica de México, 1954-05

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