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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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no 2

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 $23 \text{ Fe}^{2+} + \text{Co} \rightarrow \text{Co}^{2+} + \text{Fe}$  oxidation  
 $\text{Co} + \text{Co}^{2+} + 2\text{e}^-$  reduction  
 $\text{Fe}^{2+} + 24 \text{ Ag} \rightarrow \text{Ni}^{3+} + 3 \text{ Ag}$  oxidation  
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 $25 \text{ Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$  oxidation  
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 $\text{a} \text{ I}_2\text{O}_5 + 5\text{Co} \rightarrow \text{I}_2 + 5\text{Co}_2\text{O}_3$   
 $\text{b} \text{ O}_2 + 5\text{C} \rightarrow \text{CO}_2$

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