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molality $\frac{\text{moles solute}}{\text{kilograms solvent}}$
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web 1 molarity m mass molar mass $\times 0.4500$ 10.825 g 141.9579 g/mol $\times 0.0129 \text{ m}$
2 molality 0.825 g 141.9579 g/mol 0.00581158 mol 0.00581158 mol 0.4500 kg 0.0129 m
3 mole fraction n_2 h_2 0.825 g 141.9579 g/mol 0.00581158 mol
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begin array $\frac{\text{mass of a component}}{\text{mass of component in solution total mass of solution}} \times 100$ end array
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