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number of mol l of h<sub>2</sub> or i<sub>2</sub> that is lost to reach equilibrium  
h<sub>2</sub> g i<sub>2</sub> g 2 hi g

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n<sub>2</sub> g 3h<sub>2</sub> g 2nh<sub>3</sub> i write the initial concentrations of

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mol l i<sub>2</sub> 1 49 x 10<sup>-3</sup> mol l

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m 1 000 m 0 400 m trial 5 a g b g c g trial 6 a g b g c g initial  
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