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a 4 nh<sub>3</sub> + 5 o<sub>2</sub> → 2 n<sub>2</sub> + 6 h<sub>2</sub>o  
b h<sub>2</sub> + o<sub>2</sub> → 2 hf  
c fe<sub>2</sub>o<sub>3</sub> + 3 co → 2 fe + 3 co<sub>2</sub>  
d 2 h<sub>2</sub> + o<sub>2</sub> → 2 h<sub>2</sub>o  
answer a here ammonia nh<sub>3</sub> is the reducing agent b here water h<sub>2</sub>o is the reducing agent

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