

# Manual De Anestesia Local

**Jin-Ying Zhang**

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balance the equations below  
 $1 \text{ n}_2 + 3 \text{ h}_2 \rightarrow 2 \text{ nh}_3$   
 $3 \text{ kclo}_3 \rightarrow 3 \text{ kcl} + 3 \text{ o}_2$   
 $2 \text{ nacl} + 1 \text{ f}_2 \rightarrow 2 \text{ naf}$   
 $1 \text{ cl}_2 + 4 \text{ h}_2 \rightarrow 2 \text{ h}_2\text{o} + 2 \text{ h}_2$   
 $2 \text{ hcl} + 2 \text{ h}_2\text{o} + 1 \text{ pbcl}_2 \rightarrow 6 \text{ albr} + 3 \text{ k}_2\text{so}_4 + 6 \text{ kbr} + 1 \text{ al}_2\text{so}_4 + 3 \text{ h}_2 + 1 \text{ ch}_4 + 2 \text{ o}_2 + 1 \text{ co}_2 + 2 \text{ h}_2\text{o} + 8 \text{ c} + 3 \text{ h}_2 + 8 \text{ o}_2 + 3 \text{ co}_2 + 4$

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web 1 78 348 stoichiometry helps us determine how much substance is needed or is present things that can be measured are reactants and products mass molecular weight chemical equations formulas stoichiometric coefficient the stoichiometric coefficient or stoichiometric number is the number of molecules that participate in the reaction  
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web expert answer transcribed image text balancing equations and simple stoichiometry balance the following equations  
 $1 \text{ n}_2 + \text{f}_2 \rightarrow \text{nfs}$   
 $2 \text{ c}_6\text{h}_{10} + \text{o}_2 \rightarrow \text{co}_2 + \text{h}_2\text{o}$   
 $3 \text{ hbr} + \text{khco}_3 \rightarrow \text{h}_2\text{o} + \text{kbr} + \text{co}_2 + \text{gabr} + \text{nabr} + \text{sno}$   
 $4 \text{ naso}_3 + \text{gaz} \rightarrow 3 \text{ so}_3 + 3 \text{ nf}_3 + \text{snf}_2 + \text{n}_2\text{o}_3$   
 using the equation from problem 2 above answer the following

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1 h<sub>2</sub>o 1 kbr 1 co<sub>2</sub> br 4 2 gabr<sub>3</sub> 3 na<sub>2</sub>so<sub>3</sub> 1 ga<sub>2</sub> so<sub>3</sub> 3 6 nabr br 5 3 sno 2 nf<sub>3</sub> 3 snf<sub>2</sub> 1 n<sub>2</sub>o<sub>3</sub> br using the following equation br 2 naoh h<sub>2</sub>so<sub>4</sub> 2 h<sub>2</sub>o na<sub>2</sub>so<sub>4</sub> br

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3 00 kg of sic answer ce sio<sub>2</sub> 3c → sic 2co

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equation text  $\text{Mg} + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$  note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem  $7.4 \times 10^6$

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