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Michael Seilmaier

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web symbol for candium or whatever you used being careful not to use a symbol that is already in use such as cd or cn for your atomic number do not use one that is already in use such as 98 use the average atomic mass that

you calculated in lab 2 see the background information section at the beginning of this lab how many neutrons each do

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but different masses due to different numbers of neutrons in the nuclei the atomic mass of an element is the weighted average of the masses of the different isotopes

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lab the atomic mass of candium

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candium 2 separate the three isotopes
and count obtain the total mass for
each group of isotopes 3 complete the
data table using the instructions found
under the data analysis questions
section the large x s in the table
indicate cells that do not require data

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web candium is an interesting element
found only at fordham high school for
the arts we will be using this special
new element today to learn about
atomic mass calculations the atomic
mass listed on the periodic table for
each element is a weighted average
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web 1 obtain a sample of candium
separate the 3 isotopes candium red
hots mini candium peanut butter
regular and candium peanut large and
record the numbers for each type of

isotope 2 next determine the mass for
each isotope and calculate the total
mass in grams 3 calculate the average
mass

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questions 1 no the average atomic mass
of my group s candium sample was very
similar to our neighbors samples if each
group were to have used larger
samples of candium the respective
differences would have been smaller
because the larger the sample size the
greater the

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your teacher 2 separate the three
isotopes m m s skittle s and peanut m m
s and measure the mass of each isotope
record in data table 1 1 3 count the
number of m m s skittle s and peanut m
m s record in data table 1 1

candium radioactive decay lab sc3 a

c d also known as candium 2 - Apr
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web procedure 1 place the candium
atoms in a paper towel or a small bag 2
gently shake the atoms for 30 seconds
this is the estimated half life of your
candy 3 gently pour out the atoms onto
a paper towel 4 count the number of
atoms with the print side up these
atoms have decayed 5

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mass of a fictitious element called
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