

Armi Acciaio E Malattie Riassunto

Chao Zhang

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which is heaviest a 25 g of hg b 2 mole

of h₂o c 2 mole of co₂ d 4 g atom of o q

2 16 g of sox occupies 5 6 litre at stp

assuming ideal gas nature the value of

x is a 1 b 2 c 3 d none of these q 3 the

density of liquid mol wt 70 is 1 2 g ml 1

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02x10²³ 5 hint first find the number of

moles of oxygen used 1027 8g 1023 2g

4 6g o₂ 2 16 32 g mol n m mr n 4 6 32

n 0 144 mol y n l y 0 144 6 02 1023 y 8

669 molecules

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unbelievably small atoms and

molecules that make something up to

the measurable properties such as

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understanding of the mole concept is essential in order to do any calculations on experimental data

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one of the topics with which you leave

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being taught earlier is that the concept

of mole will be required in almost every

other topic of physical chemistry that

you study later irrespective of the

complexity of that topic

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web we interpret this as two molecules

of hydrogen react with one molecule of

oxygen to make two molecules of water

the chemical equation is balanced as

long as the coefficients are in the ratio

2 1 2 for instance this chemical

equation is also balanced 100h₂ 50o₂

100h₂o 100 h 2 50 o 2 100 h 2 o

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concept is a convenient method of

expressing the amount of a substance

any measurement can be broken down into two parts the numerical magnitude and the units that the magnitude is expressed in for example when the mass of a ball is measured to be 2 kilograms the magnitude is 2 and the unit is

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how many atoms are contained in the

following a 1 molecule of ch₃ co 2 h b

15 molecules of nh₄ cl c 2 56 mol of nh

4 3 po 4 d 0 0871 mol of cuso₄ 5h 2 o

11 find the mass in grams of each of the

following a 1 pb atom b 235 ag atoms c

4 51 x 10²² h 2 o

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