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 $1.00 \text{ MJ } 0.900 \text{ J/g } ^\circ\text{C } 80 \text{ } ^\circ\text{C } 14 \text{ kJ}$
 $q = mc \Delta t$
 $1.00 \text{ MJ } 3.88 \text{ J/g } ^\circ\text{C } 80 \text{ } ^\circ\text{C } 3.22 \times 10^3 \text{ mg } 3.2 \text{ kg}$

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no is reacted with 10 g h 2 o

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reaction was multiplied by 2 therefore
 Δh must also be halved 3 answer c
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of o 2 and not for 2 moles even though
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 Δh for the process

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