

Follada En La Primera Cita

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web when a quadratic function is represented in vertex form following points are to be noted if $h > 0$ graph shifts right by h units if $h < 0$ graph shifts left by h units if $k > 0$ graph shifts upwards by k units if $k < 0$ graph shifts downwards by k units h

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table using s p d f notation what is the

electron configuration for a neutral

atom of beryllium choose 1 answer 1 s

$2s^2$ a $1s^2 2s^2 1s^2 2s^1 2p^1$ b $1s^2$

$2s^1 2p^1 1s^2 2p^2$ c $1s^2 2p^2 1s^2$

$2p^2 1s^2 1p^2$

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configurations of elements contains all

the elements in increasing order of

atomic number to save room the

configurations are in noble gas

shorthand this means part of the

electron configuration has been

replaced with the element symbol of

the noble gas symbol

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web elements in the second row of the

periodic table place their electrons in

the $2n$ shell as well as the $1n$ shell for

instance lithium li has three electrons

two fill the $1s$ orbital and the third is

placed in the $2s$ orbital giving an

electron configuration of $1s^2 2s^1$

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answer these discussion questions

given the following part of an electron

configuration $3s^2$ explain which part

refers to the number of electrons the

energy level and the sublevel what is

the maximum number of electrons that

can occupy a single orbital state and

explain hund s rule

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web one more point needs to be

emphasized about the relationship

between electron configuration and the

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the same vertical column of the table

have similar electron configurations for

example consider the alkaline earth

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web the electron removed during the

ionization of beryllium he $2s^2$ is an s

electron whereas the electron removed

during the ionization of boron he $2s^2$

$2p^1$ is a p electron this results in a

lower first ionization energy for boron

even though its nuclear charge is

greater by one proton

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have been used for these elements

grayed out electron numbers indicate

subshells filled to their maximum

bracketed noble gas symbols on the left

represent inner configurations that are

the same in each period written out

these are he 2 helium $1s^2$ ne 10 neon

$1s^2 2s^2 2p^6$

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