

Monster From Jekyll Island

S Ashworth

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 the concentrations of reactants to the
 rate of the reaction first order reaction
 in this reaction the reaction rate is
 directly proportional to the
 concentration of only one reactant
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 that holds together the atoms ions or
 groups of atoms in a molecule or
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electrically charged positive or negative
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will freeze if the temperature is below 0
c 3 a law states a consistently observed
phenomenon can be used for prediction
b theory a widely accepted explanation
of the behavior of matter c hypothesis a
tentative explanation can be
investigated by experimentation 5
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in a series of oxyacids with similar
formulas the higher the
electronegativity of the central atom
the stronger is the attraction of the
central atom for the electrons of the
oxygen s the stronger attraction of the
oxygen electron results in a stronger
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a solution common ion effect the
lowering of the solubility of an ionic
compound as a result of the addition of
a common ion is called this lower the
smaller the numerical value of the
solubility product constant the the
solubility of the compound precipitate
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both types of bonds result from overlap
of atomic orbitals on adjacent atoms

and contain a maximum of two electrons differences σ bonds are stronger and result from end to end overlap and all single bonds are σ bonds π bonds between the same two atoms are weaker because they result from side by side

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