

Crack Coding Interview

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order of increasing polarity c f be f o f o cl s br c p c s b f n o
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polar molecule is one where the charge distribution around
the molecule is not symmetric it results from having polar
bonds and also a molecular structure where the bond
polarities do not cancel

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electrons are shared and the bond is electronegativities
equally unequally non polar polar unequally polar the degree
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molecules and ions contain polar bonds which of these
molecules and ions have dipole moments a clf₅ b clo₂ clo₂
c tecl₂ 4 tecl₄ 2 d pcl₃ e sef₄ f ph₂ ph₂ g xef₂ answer
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web the relative electronegativities of the bonded atoms is h
c cl and so the bond moments all point toward the cl end of
the molecule and sum to yield a considerable dipole moment
the molecules are relatively polar for molecules of high
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