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such as ce hf is a polar molecule as mentioned in section 4 7 because the electrons in the bond are nearer to the f atom this side of the molecule takes on a partial negative charge which is represented by δ^- δ^- is the lowercase greek letter delta

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o q2 determine based on

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nonpolar hcn h c n cf₄ c f₄ cocl₂ c o c l₂ nf₃ n f₃ q3

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together and create temporary

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living things are made up of atoms but

in most cases those atoms aren t just

floating around individually

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the following molecules and ions

contain polar bonds which of these

molecules and ions have dipole

moments a clf 5 b clo 2 clo 2 c tecl₂ 4

tecl₄ 2 d pcl₃ 3 e sef₄ f ph₂ ph₂ g xef

2 answer problem 6 2 3 6 2 3

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the bonded atoms is h c cl and so the

bond moments all point toward the cl

end of the molecule and sum to yield a

considerable dipole moment the

molecules are relatively polar for

molecules of high symmetry such as bf₃

trigonal planar ch₄ tetrahedral pf₅

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are unequally distributed