

Grupo 13 Tabla Periodica

Ensheng Dong

Principios de Química. La búsqueda del conocimiento. P. W. Atkins, Loretta Jones, 2006

Química e investigación criminal Matthew E. Johll, 2008 La fascinación de la sociedad por las investigaciones criminales se refleja en la popularidad de las novelas y series de televisión de gran audiencia basadas en la resolución de crímenes y el trabajo policial. Las historias de Sherlock Colmes, las novelas de misterio de Agatha Christie, las aventuras de investigadores inolvidables como Nancy Drew, Dick Tracy, Perry Mason y Colombo, y últimamente, la exitosa serie de televisión CSI: Crime Scene Investigation, que intriga a millones de personas, son claros ejemplos. El profesor Matthew E. Johll ha encontrado en este interés por la ciencia forense una excelente oportunidad para escribir su libro Química e investigación criminal con la finalidad de atraer a los estudiantes hacia el estudio de la química. Así, basándose en casos reales de la ciencia forense y la investigación criminal, explica los principios químicos básicos de una manera clara y accesible, sin olvidar los aspectos más cuantitativos de la química.

Química 2. Química en acción Sara Aldabe, Pedro Aramendia,

La tabla periódica Eric Scerri, 2025-01-16 Piedra Rosetta de la naturaleza, punto de partida de toda la Química, la tabla periódica de los elementos no es solo la compilación de conocimientos más compacta y significativa elaborada hasta hoy, sino que refleja el orden natural de las cosas en el mundo y, por lo que sabemos, en todo el universo. En este libro, Eric Scerri relata los antecedentes y concepción de este instrumento maravilloso, a qué obedece su lógica interna, cuál es su lugar dentro de la historia de la ciencia y de la química, y quiénes fueron y son sus principales hacedores y protagonistas. Segunda edición actualizada

Análisis en la formación de nuevas sustancias Torres González, Fidel Ángel, 2025-10-07 Texto donde se abordan temas relacionados con la identificación de los diferentes tipos de reacciones químicas de acuerdo con la subestructura atómica, así como la energía que se absorbe y/o se libera; la ubicación de los elementos químicos en la tabla periódica de acuerdo con la configuración electrónica; la identificación de la atracción de los cationes y los aniones para formar nuevos compuestos, la transferencia de materia y energía, considerando el equilibrio en las reacciones químicas y nucleares, y el impacto ambiental de las reacciones nucleares.

McGraw-Hill Education Preparación para el Examen TASC Kathy A. Zahler, 2016-03-25 The only official guide to the

SPANISH VERSION of the new high school equivalency test! The TASC high school equivalency test is given in both English and Spanish. Students who pass the Spanish-language version earn a valid high school equivalency diploma. McGraw-Hill Preparacion para el Examen TASC walks readers step-by-step through the Spanish-language version of the test. It features: a pre-test in every section that helps readers identify strengths and weaknesses before study topic-by-topic review of all test subjects and exercises to reinforce new skills full-length practice TASC test sections in the latest test format, with complete explanations for every question. For 2015 and beyond, the test is evolving toward greater rigor in the application of the Common Core State Standards and the use of new technology based question types – “hot spot” questions, “drop-down” questions, and “drag-and-drop” questions. All of these features are incorporated into Preparacion para el Examen TASC.

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Holt química ,2004

Guía bibliográfica analítico-temática de ciencias naturales Elena Ruiz de Barrios Klée,1974

Materia Bloque Fundamental Del Universo Maton,1994

Avance y perspectiva ,2001

Introducción a la química Leo J. Malone,2006 Química : materia, cambios y energía - Mediciones en química - El átomo, la estructura de la materia y las reacciones nucleares - El arreglo de los electrones en el átomo - La naturaleza periódica de los elementos - La naturaleza del enlace químico - La nomenclatura de los compuestos - Relaciones cuantitativas : la mol - Reacciones químicas : ecuaciones y cantidades - El estado gaseoso de la materia - Agua : los estados líquidos y sólidos - Soluciones acuosas - Acidos, bases y sales - Reacciones de oxidación-reducción - Cinética y equilibrio químico - Introducción a la química orgánica.

Interacciones químicas McDougal Littell,2004-09-07

Cosmos Carlos Herrero Bervera,2009

Diccionario de mecánica y estructuras metálicas Juan José Fernández,2009-02-10 Diccionario con términos de

mecánica y estructuras metálicas. Libro técnico. Pertenece a la colección Monografías de la Construcción. El Diccionario de mecánica y estructuras metálicas es una obra de referencia útil y rigurosa, una herramienta de consulta actualizada en el sector de mecánica y construcciones metálicas, apto para especialistas y profanos en el sector. Los criterios que se han seguido en la confección de esta obra han sido los siguientes:- los términos serán los específicos y más significativos del sector- los términos se corresponden con conceptos, procedimientos, actitudes profesionales, así como las técnicas más representativas, equipos, útiles y herramientas- se resuelven las principales dudas técnicas y conceptuales relacionadas con el sector de la mecánica- las definiciones permiten conocer de forma exacta los significados de los elementos tangibles y no tangibles del sector de la mecánica.

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Cationes, su análisis químico, microquímico, microscópico y espectrográfico Julio Muhana,1976

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