

# Memorie Di Un Pazzo Flaubert

**KJ Lindholm-Leary**

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web this is like a one dimensional system whose mechanical energy  $e$  is a constant and whose potential energy with respect to zero energy at zero displacement from the spring  $s$  unstretched length  $x_0$  is  $u = \frac{1}{2} kx^2$  figure 8 5 2 8 5 2 a glider between springs on an air track is an example of a horizontal mass spring system

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web 1 if the  $pe$  of the reactants is 430kJ and the  $pe$  of the products is 190kJ what is the  $\Delta h$  2 if the  $pe$  of the reactants is 430kJ and the activation energy is 220 how much total energy will be needed to be added for the reaction to

proceed transcribed image text potential energy diagram  
answer the questions using the graph

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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and  $\Delta h$  is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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web answer see below explanation the rate of a chemical reaction a b c d is given in first approximation by  $v = k[A]^m[B]^n$  where  $k$  depends by many factors as solvent geometry type of reaction  $E_a$  is the activation energy  $R$  is the universal constant of gases  $T$  is the absolute temperature

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate  $\Delta h$  for the reaction and label it with the correct value see attached image

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what effect adding a catalyst or inhibitor has on the diagram

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y axis and the x axis indicates geometry energy is a function

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has the same molecular formula same atoms and electrons

the bonding can change however

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reaction recall that the enthalpy change  $\Delta h$  is positive for an

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web 1 draw a potential energy diagram for an endothermic

reaction 2 label your drawing with the following letters a pe

of products d energy of activation b pe of reactants e heat of

reaction c pe of the activated complex potential energy

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